New Steroidal Alkaloids from Buxus longifolia

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Received April 16, 1997®

Four new steroidal alkaloids, (+)-cyclovirobuxeine F (1), *N*-benzoyl-*O*-acetylbuxalongifoline (2), buxasamarine (3), and (+)-cyclobuxamidine (4), along with two known steroidal bases, 16α -acetoxybuxabenzamidienine (5) and *trans*-cyclosuffrobuxinine (6), were isolated from the leaves of *Buxus longifolia*. The alkaloids 1-4 showed significant antibacterial activity.

Extracts of Buxus species have been used in the indigenous system of medicine for the treatment of a number of diseases including skin diseases, rheumatism, and malaria.¹ Phytochemical investigations on various Buxus species have resulted in the isolation of a large number of steroidal bases.²⁻⁶ Our previous work on the leaves of Buxus longifolia Boiss (Buxaceae) of Turkish origin has resulted in the isolation of three new steroidal alkaloids.^{7,8} Herein we describe the isolation and structure determination of four new steroidal bases from the same plant source, (+)-cyclovirobuxeine F $[(20.S)-16\alpha$ -acetoxy-3 β -(benzoylamino)-20-(dimethylamino)-31,6 α -epoxy-9,10-*seco*-buxa-9(11),10(19)-diene] (1); *N*-benzoyl-*O*-acetylbuxalongifoline [(20*S*)-16α-acetoxy-3 β -(benzoylamino)-20-(dimethylamino)-31-oxo-9,10-secobuxa-9(11),10(19)-diene] (2); buxasamarine $[(20S)-16\alpha$ hydroxy- 3β -(dimethylamino)-20-(methylamino)buxane] (3); and (+)-cyclobuxamidine $[(20.S)-3\beta-(N-1))$ methylacetylamino)-20-(methylamino)-31-norbuxa-4(30)ene] (4), along with two known steroidal bases, 16α acetoxybuxabenzamidienine $[(20.S)-16\alpha-acetoxy-3\beta-$ (benozylamino)-20-(dimethylamino)-9,10-seco-buxa-9(11),-10(9)-diene] (5) and *trans*-cyclosuffrobuxinine $[3\beta$ -(methylamino)-16-oxo-31-norbuxa-4(30),17(20)(cisoid)-diene] (6), from leaves of *B. longifolia*. Alkaloids 5 and 6 are reported for the first time from this plant.

Results and Discussion

The EtOH extract of the leaves of *B. longifolia*, when processed by column chromatography and preparative TLC, led to the isolation of four new alkaloids, (+)-cyclovirobuxeine F (1), *N*-benzoyl-*O*-acetylbuxalongifoline (2), buxasamarine (3), and (+)-cyclobuxamidine (4).

(+)-Cyclovirobuxeine F (1) had the molecular composition $C_{35}H_{48}N_2O_4$ as inferred from the HREIMS, which exhibited the M⁺ at m/z at 560.3531 (calcd 560.3539), indicating 13 double-bond equivalents in the molecule. The MS gave an ion at m/z 545 resulting from loss of a CH₃ group from the M⁺. Ions at m/z 171 and 157 were due to cleavage of ring D along with the nitrogencontaining side chain at C-17. These ions suggested that C-16 is the probable position of the acetoxy substituent on ring D.⁴ The base peak at m/z 72 represents the trimethyliminium cation, $H_3C-CH=N^+(CH_3)_2$. An ion at m/z 105 is consistent with a Ph-C=O⁺ fragment.

The ¹H-NMR spectrum of **1** showed three tertiary CH₃ singlets at δ 0.69, 0.73, and 1.90, while the secondary CH₃ resonated as a doublet at δ 1.41 ($J_{21,20} = 6.0$ Hz). These observations favor a cycloartenol-type triterpenoid skeleton, as described in other Buxus alkaloids.³⁻⁶ The amidic NH doublet at (δ 6.60) and 5H multiplets between δ 7.45 and 7.85 indicated a benzamidic substituent. The amidic NH showed coupling with a methine proton that resonated as a multiplet at δ 4.85 attributed to H-3, which showed vicinal couplings with H-2 (δ 1.70, 1.81), which were in turn coupled with the H-1 (δ 2.67). Two AB doublets resonating at δ 3.64 and 3.70 ($J_{31\alpha,31\beta} = 8.6$ Hz) were assigned to an oxygenbearing CH₂ (H-31). The downfield CH multiplet at δ 4.90 and a CH₃ singlet at δ 2.13 indicated an acetate function probably at C-16 on ring D. Another downfield multiplet at δ 4.86 was assigned to the CH proton of the oxygen-bearing C-6. The ethereal nature of the C-31 oxygen was initially inferred from the molecular composition, which showed four oxygen atoms. Three of these were accounted for by acetate and amide functionalities. The remaining oxygen was accounted for by an ether link between two carbon atoms that appeared downfield at δ 78.6 and 75.2. This was further confirmed by ¹H-NMR in pyridine- d_5 . The C-31 CH₂ protons showed only a slight paramagnetic shift from δ 3.64 to 3.74 and from δ 3.70 to 3.76. The C-6 CH also showed only a slight paramagnetic shift from δ 4.86 to 5.00. These weak paramagnetic shifts also indicated the presence of an ether linkage between C-31 and C-6, inasmuch as more pronounced solvent-induced downfield chemical shifts (ca. 1.0 ppm) would have been expected in case of an -OH substitutent as reported earlier.^{4,9} The H-3 (δ 4.85), showed a weak W(1,3-cisdiaxial) coupling with the H-5 (δ 2.07) in the COSY-45° spectrum. Couplings among H-5/H-6, H-6/H-7, and H-7/ H-8 were also observed in the COSY spectrum. A broad singlet that appeared at δ 5.46 was assigned to the vinylic proton (H-11), and the broad singlet at δ 5.87 was assigned to the olefinic proton (H-19). H-19 showed a weak allylic coupling with H-11 (olefinic), which in turn showed vicinal coupling with H₂-12 (δ 2.10). These observations suggested the presence of an abeo-diene system, which was confirmed by the characteristic UV absorptions.¹⁰

A broad-band decoupled ¹³C-NMR spectrum of 1

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Table 1. ¹H- and ¹³C-NMR Chemical Shift Data (CDCl₃) of Compounds 1, 2, and 3

	1		2		3	
	¹³ C δ	$^{1}\text{H}\delta$	¹³ C δ	${}^{1}H \delta$	¹³ C δ	$^{1}\text{H}\delta$
carbon	(mult, DEPT)	(mult, $J = Hz$)	(mult, DEPT)	(mult, $J = Hz$)	(mult, DEPT)	(mult, $J = Hz$)
1	37.5 (CH ₂)	2.67 (m)	39.1 (CH ₂)	1.80 (m)	127.0 (CH)	5.62 (d) $(J_{1,2} = 10.5)$
2	34.2 (CH ₂)	$H_{\alpha} = 1.70 \text{ (m)}$	25.1 (CH ₂)	2.05 (m)	128.0 (CH)	5.40 (m)
		$H_{\beta} = 1.70$ (m)		2.20 (m)		
3	56.6 (CH)	4.85 (m)	56.8 (CH)	4.30 (m)	71.3 (CH)	2.50 (m)
4	44.8 (C)		48.7 (C)		41.3 (3)	
5	42.2 (CH)	2.07 (m)	50.5 (CH)	1.40 (m)	48.7 (CH)	1.87 (m)
6	75.2 (CH)	4.86 (m)	29.7 (CH ₂)	1.24 (m)	20.0 (CH ₂)	1.50 (m)
-				1.32 (m)		
7	26.8 (CH ₂)	2.10 (m)	36.7 (CH ₂)	1.82 (m)	31.1 (CH ₂)	1.25 (m)
•				1.90 (m)		
8	47.6 (CH)	2.00 (m)	42.1 (CH)	2.00 (m)	44.1 (CH)	2.00 (m)
9	134.7 (C)		134.0 (C)		19.0 (C)	
10	133.3 (C) 197.2 (CII)	E 46 (m)	137.0 (C) 196.0 (CH)	E E A (ma)	20.0(C)	1.95 ()
11	127.3 (CH)	5.40 (III)	120.0 (CH)	5.54 (III)	24.9 (CH ₂)	1.85 (III) 1.02 (m)
19	42.7 (CL.)	2 10 (m)	27 5 (CU.)	9.15(m)	21 7 (CL)	1.92 (III) 1.70 (m)
12	43.7 (C11 ₂)	2.10 (III)	37.3 (CII ₂)	2.13 (III) 2.25 (m)	$31.7 (C11_2)$	1.70 (11)
13	46 2 (C)		44.0(C)	2.23 (III)	45 3 (C)	
14	40.2 (C)		47.0(C)		49.7 (C)	
15	32.9 (CH ₂)	1.50 (m)	42.6 (CH ₂)	2.12 (m)	42.1 (CH ₂)	2.00 (m)
10	02.0 (0112)	1.61 (m)	12.0 (0112)	2.12 (III)	12.1 (0112)	2.00 (11)
16	75.2 (CH)	4.90 (m)	79.3 (CH)	5.10 (m)	77.7 (CH)	4.24 (m)
17	53.6 (CH)	2.05 (m)	56.4 (CH)	2.03 (m)	59.8 (CH)	1.75 (m)
18	16.0 (CH ₃)	0.69 (s)	16.7 (CH ₃)	0.85 (s)	17.8 (CH ₃)	0.90 (s)
19	128.4 (CH)	5.87 (s)	128.0 (CH)	5.90 (s)	18.25 (CH ₂)	-0.11 (d) $(J_{19\alpha,19\beta} = 4.1)$
						0.70 (d) $(J_{19\beta,19\alpha} = 4.1)$
20	65.5 (CH)	3.20 (m)	65.4 (CH)	3.00 (m)	58.8 (CH)	2.85 (m)
21	12.4 (CH ₃)	1.41 (d) $(J_{21,20} = 6.0)$	11.2 (CH ₃)	1.28 (d)	15.1 (CH ₃)	1.12 (d) $(J_{21,20} = 6.1)$
				$(J_{21,20} = 6.5)$		
30	19.9 (CH ₃)	1.90 (s)	21.0 (CH ₃)	1.80 (s)	26.0 (CH ₃)	1.15 (s)
31	78.6 (CH ₂)	$H\alpha = 3.64$ (d) $(J_{31\alpha,31\beta} = 8.6)$	201.0 (CH)	9.53 (s)	15.5 (CH ₃)	1.05 (s)
		$H_{\beta} = 3.70$ (d) $(J_{31\beta,31\alpha} = 8.6)$				
32	17.5 (CH ₃)	0.73 (s)	17.6 (CH ₃)	0.84 (s)	16.5 (CH ₃)	$0.77 (CH_3)$
N _a -CH ₃	00.0 (CII.)	0.05()		0.70()	$31.7 (CH_3)$	2.27 (s)
N _b -CH ₃	36.0 (CH ₃)	2.65 (s)	46.7 (CH ₃)	2.70 (s)	32.6 (CH ₃)	2.40 (s)
N_b-CH_3	43.4 (CH ₃)	2.65 (S)	127 0 (C)			
C^{-1}	132.0 (C) 199.7 (CH)	7.95 (m)	137.0 (C) 128.0 (CH)	7.71 (m)		
C-2 C 2'	120.7 (CH)	7.65 (III) 7.45 (m)	120.0 (CH)	7.71 (III) 7.40 (m)		
C-1'	130.1 (CH) 131.0 (CH)	7.45 (III) 7.75 (m)	129.3 (CH) 131.3 (CH)	7.40 (III) 7.46 (m)		
C-4 C-5'	131.0 (CH) 130.1 (CH)	7.75 (III) 7.45 (m)	129 3 (CH)	7.40 (m)		
C-6′	128 7 (CH)	7.85 (m)	128 0 (CH)	7 71 (m)		
Ph-CO-N	171 3 (C)		166 0 (C)			
H <u>₂C-C</u> O-	177.3 (C)		170.3 (C)			
$H_3C-\overline{CO}$	21.2 (CH ₃)	2.13 (s)	21.5 (CH ₃)	2.07 (s)		
	(· · · · · · · · · · · · · · · · · · ·	(3)			

showed 35 carbon resonances. The DEPT spectra indicated that there were seven CH_3 , six CH_2 , fourteen CH, and (by difference from the broad-band decoupled spectrum) eight quaternary carbon atoms. Signals for the vinylic C-11 and C-19 appeared at δ 127.3 and 128.4. Two downfield quaternary signals at δ 171.3 and 177.3 were assigned to the benzamidic and ester carbonyl carbons. One-bond ¹H⁻¹³C correlations were established by heteronuclear multiple quantum coherence (HMQC) NMR experiments.¹¹⁻¹⁴ The chemical shifts and various proton–carbon connectivities are given in Table 1.

The long-range ${}^{1}\text{H}{-}{}^{13}\text{C}$ connectivities were determined from an HMBC spectrum. 12,14 A CH carbon resonating at δ 56.6 (C-3) was coupled to protons resonating at δ 3.64 and 3.70 (H-31 α and β) and 1.70 (H-2 α). The C-4 quaternary carbon resonating the δ 44.8 exhibited couplings with the protons resonating at δ 4.85 (H-3) and 2.07 (H-5), confirming that C-3 is linked to C-4. Similarly, the olefinic C-10 (δ 133.5) exhibited couplings with the protons resonating at δ 2.67 (H-1) and 2.07 (H-5). The quaternary carbon at δ 46.2 (C-13) exhibited heteronuclear interaction with the protons

at δ 2.10 (H₂-12). The acetoxy-bearing C-16 CH proton (δ 4.90) showed ²J_{CH} couplings with C-17 (δ 53.6) and C-15 (δ 32.9), and ³J coupling with C-20 (δ 65.5) indicated that the acetoxy group was present at the C-16 position. ¹H⁻¹³C long-range connectivities in **1** are summarized in Table 2.

NOE difference NMR measurements were carried out to determine the relative stereochemistry at various asymmetric centers in **1**. For example when H-8 β (δ 2.00) was irradiated, an 8% NOE was observed at δ 4.86 (H-6), indicating the β -orientation of H-6, and therefore, C-6 ether linkage is α -oriented. When H-31 (δ 3.70) was irradiated, an 8% NOE was observed on H-30 (δ 1.90), which is β -oriented; therefore, H-31 was inferred to be α -oriented. Similarly, when H-3 (δ 4.85) was irradiated, a 14.2% NOE was observed at δ 2.07 (C-5 α) and a 5.2% NOE at δ 1.70 (equatorial H-2 α), indicating α - (pseudo axial) stereochemistry of the C-3 proton. Irradiation of H-16 (δ 4.90) resulted in a 6% NOE at δ 0.69 (β CH₃-18) and no NOE effect on the α -oriented C-17 proton, indicating that the proton at C-16 was β -oriented and that the acetoxy substituent was α -oriented. Other

1			3		
¹ Η (δ)	$^{2}J_{HC}(\delta)$	$^{3}J_{HC}(\delta)$	¹ Η (δ)	² J _{HC} (δ)	³ J _{HC} (δ)
H-2', H6'		171.3 (amide, C=O)			
(7.85)					
		131.0 (C-4')	H-1 (5.62)	20.0 (C-10)	48.7 (C-5)
H3′	131.0 (C-4')		H-2 (5.40)	127.0 (C-1)	20.0 (C-10)
(7.75)				71.33 (C-3)	
H-3′,5′	128.7 (C-2',C-6')	132.0 (C-1')	H-16 (4.24)	59.8 (C-17)	45.35 (C-13)
(7.75)				42.1 (C-15)	49.7 (C-14)
H-11	43.7 (C-12)	47.6 (C-8)	H-3 (2.50)	128.0 (C-2)	127.0 (C-1)
(5.46)				41.3 (C-4)	
H-16	32.9 (C-15)	65.5 (C-20)		. ,	
(4.90)	53.6 (C-17)				
H-3	44.8 (C-4)	78.6 (C-31)	H-8 (2.00)	19.0 (C-9)	45.3 (C-13)
(4.85)		171.3 (amide, C=O)		49.7 (C-14)	. ,
				31.1 (C-7)	
H-31 βH	44.8 (C-4)	19.9 (C-30)	H-5 (1.87)	20.0 (C-6)	31.1 (C-7)
(3.70)		56.6 (C-3)		41.3 (C-4)	127.0 (C-1)
		75.2 (C-6)			
Η-31 αΗ	44.8 (C-4)	56.6 (C-3)	H-11 (1.85)	31.7 (C-12)	18.2 (C-19)
(3.64)		42.2 (C-5)		19.0 (C-9)	
		75.2 (C-6)			
H-1	133.5 (C-10)				
(2.67)	34.2 (C-2)				
H-12	46.2 (C-13)		H-12 (1.70)	45.3 (C-13)	59.8 (C-17)
(2.10)					
H-5	44.8 (C-4)		H-7 (1.25)	44.1 (C-8)	48.7 (C-5)
(2.07)	133.5 (C-10)			20.0 (C-6)	
H-8	48.8 (C-14)		H-30 (1.15)	41.3 (C-4)	15.5 (C-31)
(2.00)					71.3 (C-3)
Η-2α	56.6 (C-3)		H-31 (1.05)	41.3 (C-4)	48.7 (C-5)
(1.70)					



Figure 1. NOE connectivities observed in 1.

important NOE interactions are delineated around Figure 1.

The second compound, N-benzoyl-O-acetylbuxalongifoline (2), was found to be similar to (+)-buxanaldinine present in the literature.⁴ The M^+ at m/z 546.3461 (calcd 546.3455) in HREIMS, corresponding to molecular formula C₃₄H₄₆N₂O₄, and indicating 13 double-bond equivalents in the molecule. The mass fragmentation pattern of 2 was distinctly similar to the known alkaloid 16α -acetoxybuxabenzamidienine (5). Both compounds gave a fragment ion at m/z 105 due to cleavage of a benozyl group. Compound 2 has an N_b-CH₃ group at C-20 position, while 5 has a N_b-(CH₃)₂ group. Thus, the fragment ions obtained from cleavage of the C-17/C-20 bond have different m/z values. Compound **2** gave the base peak at m/z 58, due to the dimethyliminium ion $H_3C-CH=+NHCH_3$, while **5** gave the base peak at m/z72 due to the trimethyliminium ion $H_3\dot{C}$ -CH=N⁺-(CH₃)₂. Similarly, compound 2 yielded two fragment ions at m/z 143 and 157, due to the cleavage of ring D, which indicated an acetate group at C-16. Compound **5**, on the other hand gave two fragment ions at m/z 157



and 171 due to the similar cleavages of ring D. The only difference in the composition of the fragment ions was the lack of one CH_3 group in **2**. The ¹H-NMR spectrum of **2** was also found to be distinctly similar to **5**, with

Table 3. Antibacterial Activity of	of Compounds 1	l, 2 ,	3 and	4
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		zone of inhibition in nm		
test bacteria	compound	100 μg/100 μL	200 μg/100 μL	MIC
Salmonella typhi	1	7	8	100 μg/100 μL
	2	6	8	$100 \mu g/100 \mu L$
	3	6	8	$100 \mu g/100 \mu L$
	4	6	6	$150 \mu g / 100 \mu L$
	ampicillin ^a	10	11	$30 \mu g/100 \mu L$
Shigella flexneri	1	7	7	$100 \mu g / 100 \mu L$
	2	6	6	200 μg/100 μL
	3		7	$150 \mu g / 100 \mu L$
	ampicillin	11	11	30 μg/100 μL
Pseudomonas aeruginosa	1	6	8	$10 \mu g / 100 \mu L$
-	2	8	9	50 μg/100 μL
	3	7	9	50 μg/100 μL
	ampicillin	11	11	40 $\mu g/100 \mu L$
Escherichia coli	4		6	200 μg/100 μL
	ampicillin	11	12	$50 \mu\text{g}/100 \mu\text{L}$

^a Ampicillin was used as standard drug.

the difference that the ¹H-NMR spectrum of **2** showed the presence of an aldehydic group (δ 9.53), most likely at C-4,⁴ and lack of one N-CH₃.

¹³C-NMR spectra of **2** showed resonances for all 34 carbons, including six CH₃, six CH₂, 14 CH, and eight quaternary carbon atoms. There were three signals of carbonyl carbons, that is, the carbon of acetyl carbonyl appeared at δ 170.3, amide carbonyl at δ 166.0, and aldehyde carbonyl carbon at δ 201.0. The chemical shifts of various carbons and their one-bond heteronuclear connectivities are presented in Table 1. The above spectroscopic studies led to structure for this new steroidal base, *N*-benzoyl-*O*-acetylbuxalongifoline.

A HREIMS of **3** showed the M⁺ at m/z 414.3600 (calcd 414.3609) corresponding to the molecular formula C₂₇H₄₆N₂O, indicating six double-bond equivalents in the molecule. The base peak at m/z 58 was attributed to the fragment ion, H₃C-CH=⁺NC CH₃. Two fragment ions at m/z 101 and 115 indicated that the OH group was at C-16 in ring D.⁵

The ¹H-NMR spectrum of **3** revealed four tertiary CH₃ groups at δ 0.77, 0.90, 1.05, and 1.15, while the secondary CH₃ resonated as a doublet at δ 1.12 ($J_{21,20}$ = 6.1 Hz). A doublet at δ 5.62 was assigned to the vinylic proton (H-1), and it showed coupling with the H-2 vinylic proton (δ 5.40); H-3 (δ 2.50) showed vicinal coupling with H-2 (δ 5.40) and W (1,3-*cis*-diaxial) coupling with H-5 (δ 1.87). The H-19H α resonated as a very upfield doublet at δ -0.11 and showed geminal coupling with H-19 β . Geminal couplings with H₂-12 (δ 1.85, 1.92) and their vicinal couplings with H₂-12 (δ 1.70) were also observed. A multiplet at δ 4.24 was assigned to the H-16, which showed couplings with the H-17 (δ 1.75) and H-15 (δ 2.00).

DEPT and broad-band decoupled ¹³C-NMR spectra of **3** showed resonances of all eight CH₃, six CH₂, eight CH, and five quaternary carbon atoms. A characteristic feature of the broad-band ¹³C-NMR spectrum was the appearance of downfield olefinic signals at δ 127.0 (C-1) and 128.0 (C-2). The ¹³C-NMR chemical shifts and one-bond heteronuclear correlations (HMQC) of **3** are presented in Table 1. Table 2 summarizes HMBC interactions.

The spectral data of (+)-cyclobuxamidine (**4**) was similar to that reported for (+)-*N*-acetyl-*N*-demethyl-cyclomicrobuxeine.¹⁵ The HREIMS of **4** exhibited M⁺ at m/z 412.3129, corresponding to the molecular formula

C₂₇H₄₄N₂O (calcd 412.3119) and indicating seven doublebond equivalents in the molecule. The base peak at m/z58 represented the dimethyliminium ion H₃C-CH=N⁺-HCH₃). Ions at m/z 85 and 99 were due to cleavage of the C-13,17/C-15,16 and C-13,17/C-14,15 bonds of ring D along with the nitrogen-containing side chain.

The ¹H-NMR spectrum of **4** revealed two tertiary CH₃ groups as two sharp 3H singlets at δ 1.20 and 0.87. The H-21 proton resonated as a doublet at δ 0.98 ($J_{21,20}$ = 6.0 Hz). The appearace of two AB doublets at δ 1.02 and 0.30 ($J_{19\alpha,19\beta} = 4.3$ Hz) was characteristic of H-19 cyclopropyl protons. H-3 appeared as a multiplet at δ 4.45 and showed W (1,3-cis-diaxial coupling with H-5 proton in the COSY-45° spectrum. The ¹H-NMR spectrum of 4 displayed doubling of many peaks, each integrating for nearly half of the actual integration. For instance, N_a -CH₃ appeared as singlets at δ 2.99 and 3.10. Similarly, the methyl of N_aCOCH₃ showed two singlets of δ 2.01 and 2.15. The exomethylene protons (H-30) also showed doubling of signals and resonated at δ 4.56/4.60 and 4.65/4.70. It was suspected that this doubling of signals in the spectrum of **4** was due to its existence of two conformers (amidic rotamers)¹⁵ at room temperature. Stereochemical assignments at the various asymmetric centers on the skeleton were made on biosynthetic considerations and are in agreement with the known *Buxus* alkaloids.¹⁶ The above spectroscopic studies led to structure 4 for this new steroidal alkaloid, (+)-cyclobuxamidine.

Two known compounds 16α -acetoxybuxabenzamidienine (**5**)⁴ and *trans*-cyclosuffrobuxinine (**6**)¹⁶ were isolated for the first time from the leaves of *B. longifolia*. These compounds were earlier isolated from *Buxus sempervirens*.

Compounds **1**–**3** showed antibacterial activity against *Salmonella typhi, Shigella flexneri,* and *Pseudomonas aeruginosa,* while compound **4** was active against *Salmonella typhi* and *Escherichia coli* (Table 3) when assayed by the tube broth dilution method.

Experimental Section

General Experimental Procedures. The MS spectra were recorded on a Varian MAT 312 double-focusing spectrometer. ¹H- and ¹³C-NMR spectra were recorded in $CDCl_3$ on a Bruker AM 400 NMR spectrometer. UV spectra were recorded on a Shimadzu UV-240 instrument. The IR spectra were recorded on a JASCO IRA-1



IR spectrophotometer. Specific rotations were measured on a Polartronic D polarimeter. Purities of the samples were checked on precoated TLC (Si gel, G₂₅₄) plates.

Plant Material. Leaves of Buxus longifoia Boiss (10 kg) were collected in Hatay-Antakya in southeastern Turkey, near St. Peter's Church, in June 1990. The plant was identified by one of us (B.S.), and a voucher specimen was deposited in the herbarium of the Faculty of Pharmacy, Gazi University, Ankara, Turkey.

Extraction and Isolation. An EtOH extract (818 g) of *B. longifolia* leaves was evaporated to a gum and redissolved in H₂O. Partial separation of the alkaloids was carried out by extraction with CHCl₃ at different pH values. The CHCl₃ extract obtained at pH 5.0 (12 g) was loaded onto a Si gel column (70–230 mesh, 360 g), which was eluted with mixture of petroleum ether-CHCl₃ in order of increasing polarity. The fraction obtained by eluting the column with petroleum ether- $CHCl_3$ (1.5:8.5) was found to contain three major compounds, by TLC, which were further purified by preparative TLC (Si gel) using petroleum ether-CHCl₃-Et₂NH (8:1.8:0.2) to afford compounds **1**, **3**, and **4**.

The CHCl₃ extract of pH 7.0 (35.2 g) was also loaded on a Si gel column (70-230 mesh, 1 kg), which was eluted with increasing polarities of mixtures of petroleum ether-CHCl₃. The fraction obtained by eluting the column with petroleum ether- $CHCl_3$ (6:4) was further purified by preparative TLC (Si gel) in petroleum ether-CHCl₃-Et₂NH (8:1.8:0.2) to give 2 along with two known steroidal bases, (+)-16 α -acetoxybuxabenzamidienine (5) and *trans*-cyclosuffrobuxinine (6).

(+)-Cyclovirobuxeine F (1): white amorphous solid; $[\alpha]^{22}_{D}$ +7° (*c* 0.11, CHCl₃); UV (MeOH) λ_{max} (log ϵ) 254 (4.35), 246 (4.40), 238 (4.32), 230 (4.31) nm; IR (CHCl₃) vmax 3406, 2813, 1712, 1610, 1358 cm⁻¹; ¹H-NMR (CDCl₃, 400 MHz) and ¹³C-NMR (CDCl₃, 100 MHz) data, see Table 1; EIMS *m*/*z* 560 [M⁺], 545 (10), 171 (11), 157 (11.4), 72 (100), 105 (23.5); HREIMS m/z 560.3531 (calcd for C₃₅H₄₈N₂O₄, 560.3539).

N-Benzoyl-O-acetylbuxalongifoline (2): white amorphous solid; $[\alpha]^{22}_{D}$ +16° (c 0.065, CHCl₃); UV (MeOH) λ_{max} (log ϵ) 254 (sh) (4.32), 245 (4.30), 235 (4.29), 229 (4.28) nm; IR (CHCl₃) v_{max} 3348, 2812, 1718, 1664,

1626 cm⁻¹; ¹H- (CDCl₃, 400 MHz) and ¹³C-NMR (CDCl₃, 100 MHz) data, see Table 1; EIMS *m*/*z* 546 [M⁺] (10), 531 (8), 58 (100), 144 (10), 157 (25), 105 (32); HREIMS m/z 546.3461 (calcd for C₃₄H₄₆N₂O₄, 560.3455).

Buxasamarine (3): white amorphous solid; $[\alpha]^{22}$ _D $+23^{\circ}$ (c 0.13, CHCl₃); IR (CHCl₃) ν_{max} 3456, 2902, 1362 cm⁻¹; ¹H- (CDCl₃, 400 MHz) and ¹³C-NMR (CDCl₃, 100 MHz) data, see Table 1; EIMS m/z 414 [M⁺] (15), 399 (10), 58 (100), 101 (12), 115 (15); HREIMS m/z 414.600 (calcd for C₂₇H₄₆N₂O, 414.3609).

(+)-Cyclobuxamidine (4): white amorphous solid; $[\alpha]^{22}_{D}$ +24° (c 0.041, CHCl₃); UV (MeOH) λ_{max} (log ϵ) 220 (4.36) nm; IR (CHCl_3) $\nu_{\rm max}$ 3673, 2918, 1685, 1607 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 0.87 (3H, s, tert-Me), 0.98 (3H, d, $J_{21,20} = 6.0$ Hz, Me-21), 1.20 (3H, s, tert-Me), 2.01, 2.15 (3H, s, COCH₃), 2.73 (3H, s, CH₃-N_b), 2.99, 3.10 (3H, s, CH₃-N_a), 4.45 (1H, m, H-3), 4.56, 4.60 (1H, s, H-30), 4.65, 4.70 (1H, s, H-30); EIMS m/z 412 [M⁺] (12), 99 (3), 113 (2), 58 (100); HREIMS m/z 412.3129 (calcd for C₂₇H₄₄N₂O, 412.3119).

(+)-16α-Acetoxybuxabenzamidienine (5): white amorphous solid; $[\alpha]^{22}_D$ +8° (c 0.015, CHCl₃); UV (MeOH) λ_{max} (log ϵ) 254 (4.40), 245 (4.35), 237 (4.32), 230 (4.30), 225 (4.28) nm; IR (CHCl₃) v_{max} 3520, 1720, 1650, 1605 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 0.75 (3H, s, tert-Me), 0.84 (3H, s, tert-Me), 0.85 (3H, s, tert-Me), 0.90 (3H, d, *J*_{21,20} = 6.5 Hz, Me-21) 1.85 (3H, s, COCH₃), 2.30 [6H, s, N, (CH₃)₂], 2.80 (1H, m, H-20), 4.35 (1H, m, H-3), 5.06 (1H, m, H-16), 5.45 (1H, s, H-11), 5.90 (1H, m, H-19), 6.50 (1H, d, $J_{\rm NH,3} = 8.2$ Hz), 7.72–7.30 (5H, m, Ar-H); EIMS 546 [M⁺] (25), 531 (18), 171 (45), 157 (50), 105 (25), 72 (100); HREIMS m/z 546.3467 (calcd for C₃₅H₅₀N₂O₃, 545.63451).

trans-Cyclosuffrobuxinine (6): white amorphous solid; $[\alpha]^{22}D - 42^{\circ}$ (c 0.012, CHCl₃); UV (MeOH) λ_{max} (log *ϵ*) 243 (3.90) nm; IR (CHCl₃) *ν*_{max} 3450, 3040, 1734, 1730, 1680, 1660, 1465 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 0.60, 0.91 (each 1H, d, $J_{19\alpha,19\beta} = 4.2$ Hz, H-19), 1.12 (3H, s, tert-Me), 1.21 (3H, s, tert-Me), 1.82 (3H, d, $J_{21,20} =$ 6.2 Hz, Me-21) 4.84 and 4.86 (each 1H, s, C=CH₂), 5.50 (1H, q, H-20); EIMS 353 $[M^+]$ (20), 383 (15), 325 (30), 70 (40), 57 (100); HREIMS m/z 353.2758 (calcd for C₂₄H₃₅NO, 353.2718).

Bioassay Procedure. Pure compounds were assaved for antibacterial activity against Salmonella typhi, Shigella flexneri, Escherichia coli, and Pseudomonas aeruginosa. The minimum inhibitory concentration (MIC) values were determined by the tube broth dilution method¹⁷ using nutrient broth. Sensitive bacterial cultures (0.1 mL of 10^4-10^6 cells/mL) were added to the tubes containing 1 mL of different concentrations of the test sample, and 1 mL of nutrient broth. A similar procedure was followed for the standard antibiotic ampicillin. Inhibition of growth was observed after 24 h of incubation at 37 °C.

Acknowledgment. We thank the INFAQ Foundation for financial support to F. N. We also wish to acknowledge ONR grant (N 00014-86-G-0229) to A. R. and TWAS grant (BC 91-016) to M.I.C.

References and Notes

- (1) Cordell, G. A. Introduction to Alkaloids: A Biogenetic Ap*proach:* John Wiley & Sons: New York, 1981; pp 907–908. Atta-ur-Rahman; Ahmed, D.; Asif, E.; Jamal, S. A.; Choudhary,
- M. I.; Sener, B.; Türköz, S. Phytochemistry 1991, 30, 1295-1298.

- (3) Atta-ur-Rahman; Iqbal, Z.; Choudhary, M. I.; Nasir, H.; Fatima,

- Atta-ur-Rahman; Iqbal, Z.; Choudhary, M. I.; Nasir, H.; Fatima, T. Phytochemistry 1990, 29, 683-685.
 Choudhary, M. I.; Atta-ur-Rahman; Freyer, A. J.; Shamma, M. Tetrahedron 1986, 42, 5747-5752.
 Atta-ur-Rahman; Ahmed, D.; Asif, E.; Ahmed, S.; Sener, B.; Türköz, S. J. Nat. Prod. 1991, 54, 79-82.
 Atta-ur-Rahman; Muzaffar, A. The Alkaloids; Brossi, A., Ed.; Academic Press: New York, 1988; Vol. 32, pp 79-233.
 Atta-ur-Rahman; Naz, S.; Noor-e-ain, F.; Ali, R. A.; Choudhary, M. I.; Sener, B.; Türköz, S. Phytochemistry 1992, 31, 2933-2935.
 Atta-ur-Rahman; Noor-e-ain, F.; Ali, R. A.; Choudhary, M. I.; Pervin, A.; Türköz, S.; Sener, B. Phytochemistry 1992, 32, 1059-1063. 1063.
- (9) Demarco, P. V.; Farkas, E.; Doddrell, D.; Mylari, B. L.; Wenkert, E. J. Am. Chem. Soc. 1968, 90, 5480-5486.
- (10) Atta-ur-Rahman; Choudhary, M. I. In *Studies in Natural Products Chemistry*, Elsevier Science: Amsterdam, 1988; Vol. 2, Part A, pp 175-185.

- (11) Atta-ur-Rahman; One- and Two-Dimensional NMR Spectroscopy, Elsevier Science: Amsterdam, 1989.
- (12) Atta-ur-Rahman; Choudhary, M. I. Solving Problems by NMR Spectroscopy; Academic Press: San Diego, 1995.
- (13) Bax, A.; Summers, M. F. J. Am. Chem. Soc. 1986, 108, 2093-2094.
- (14) Müller, L. J. Am. Chem. Soc. 1979, 101, 4481-4484.
- (15) Atta-ur-Rahman; Alam, M.; Choudhary, M. I. Phytochemistry **1988**, 27, 3342-3343.
- (16) Nakano, T.; Terao, S.; Saeki, Y. J. Chem. Soc. (C) 1966, 1412-1421.
- (17) Carran, R.; Muran, A.; Montero, J. M.; Fernandozlago, L.; Dominguez, A. Plantes Med. Phytother. 1987, 21, 195.

NP970207E